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### 12th Chemistry Solution Le9

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Mass % of benzene =  $\frac{22}{144} \times 100 = 15.28\%$  Mass % of  $\text{CCl}_4 = \frac{122}{144} \times 100 = 84.72\%$  2.2. Calculate the mole fraction of benzene in solution containing 30% by mass in carbon tetrachloride. 1.015 kg of urea solution contains 15g of urea  $\therefore$  2.5 kg of solution contains urea =  $\frac{15}{1.015} \times 2.5 = 37 \text{ g}$  (c ...

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NCERT Solutions CBSE Sample Papers Chemistry Class 12 Chemistry NCERT IN TEXT QUESTIONS. 10.1 Write structures of the following compounds: (i) 2-Chloro-3-methylpentane (ii) 1-Chloro-4-ethylcyclohexane

#### NCERT Solutions For Class 12 Chemistry Chapter 10 ...

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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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The log function also called a logarithm function which comes under most of the mathematical problems. The logarithmic function is used to reduce the complexity of the problems by reducing multiplication into addition operation and division into subtraction operation by using the properties of logarithmic functions. Here, the method to find the logarithm function is given with the help of value ...

#### Value of Log e - Log Function to the Base 10 and Base e

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